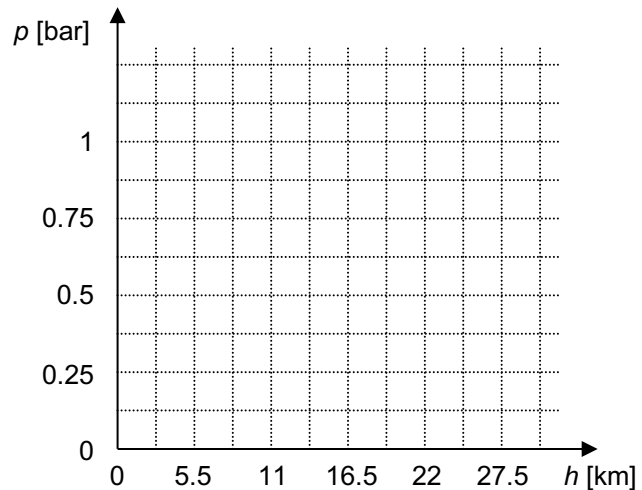
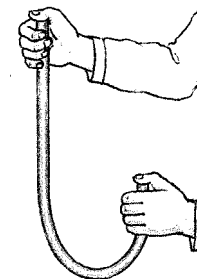


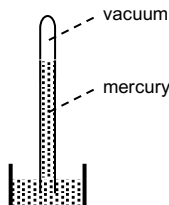
1. Atmospheric pressure and height.
  - a) What is the atmospheric pressure at sea level, 5'500 m above sea level, 11'000 m above sea level, 16'500 m above sea level, and 22'000 m above sea level?
  - b) At what height above sea level is the atmospheric pressure 15.83 mbar?
  - c) Plot the relationship between atmospheric pressure and height in the diagram to the right.
  - d) Determine from the diagram the approximate value of atmospheric pressure at 2'750 m above sea level.
  - e) Determine from the diagram the height above sea level, at which the atmospheric pressure is 375 mbar.



2. Suction cups can only be used on a very smooth surface. Why? Why does a suction cup stick to the wall? Explain without using the word *suction*.
3. Vacuum packing has a great advantage in storing food: Air is removed and bacterial growth is inhibited. The contents are sealed in a plastic bag which fits tightly to the contents. Explain why atmospheric pressure is important in vacuum packaging.
4. The Magdeburg hemispheres ( $A = 1'170 \text{ cm}^2$ ) were pressed together with a force of 11.2 kN. The atmospheric pressure was 999 mbar.
  - a) What was the gauge pressure (difference in pressure between the inside of the sphere and atmospheric pressure)?
  - b) What was the pressure on the inside of the sphere?
5. Fill a hose with water and close it at both ends with your thumbs. Then hold it as in the picture.
  - a) What happens if you open the lower end of the hose? Why?
  - b) What happens if you open the higher end of the hose (keeping the lower end open)? Why?
  - c) What would have happened on the moon (without atmospheric pressure)?



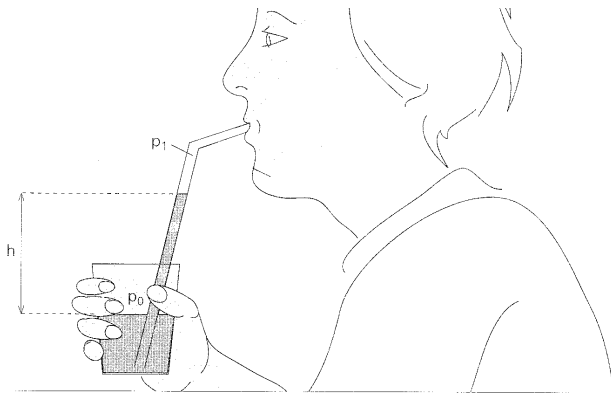
6. What is the height of the mercury column on Mount Everest ( $p = 335 \text{ mbar}$ )?



7. A syringe is easily filled with water.
  - a) Explain the process without using the word *suction*.
  - b) Would it be possible to fill a syringe with water on the moon just as easily as on earth? Give reasons for your answer.

8. Drinking water through a straw at sea level.

- Explain the process without using the word *suction*.
- What is the atmospheric pressure  $p_0$  at the surface of the water (outside the straw)?
- Calculate the fluid pressure at the surface of the water column inside the straw ( $h = 20.0$  cm).
- At the water surface the pressure must be the same inside and outside the straw. The pressure of the water column and the pressure  $p_1$  in the upper part of the straw add up to equal the atmospheric pressure  $p_0$ .  
What is the pressure  $p_1$ ?
- What is the gauge pressure in the upper part of the straw? (Difference between atmospheric pressure and pressure inside the upper part of the straw)
- Would it be possible to drink through a straw on the moon? Give reasons for your answer.




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*solutions*

- |   |  |
|---|--|
| 1. a) 1'013 mbar, 506.5 mbar, 253.3 mbar, 126.6 mbar. 63.3 mbar | b) 33'000 m above sea level                |
| d) ca. 750 mbar   | e) ca. 7'000 m above sea level             |
| 4. a) 957 mbar  | b) 42 mbar                                 |
| 6. 25 cm  |  |
| 8. b) 1.013 bar   | c) 1'958 Pa    d) 99.3 kPa    e) -1'958 Pa |