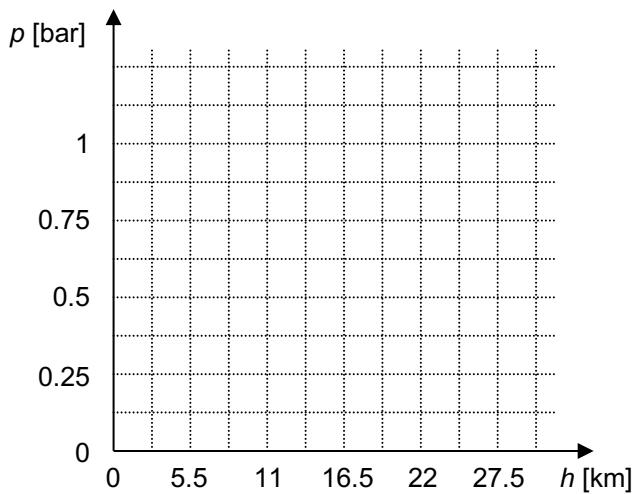
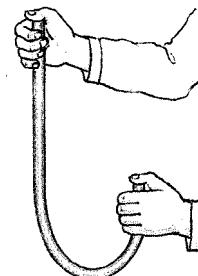
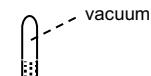


- Atmospheric pressure and height.
 - What is the atmospheric pressure at sea level, 5'500 m above sea level, 11'000 m above sea level, 16'500 m above sea level, and 22'000 m above sea level?
 - At what height above sea level is the atmospheric pressure 15.83 mbar?
 - Plot the relationship between atmospheric pressure and height in the diagram to the right.
 - Determine from the diagram the approximate value of atmospheric pressure at 2'750 m above sea level.
 - Determine from the diagram the height above sea level, at which the atmospheric pressure is 375 mbar.



- Suction cups can only be used on a very smooth surface. Why? Why does a suction cup stick to the wall? Explain without using the word *suction*.
- Vacuum packing has a great advantage in storing food: Air is removed and bacterial growth is inhibited. The contents are sealed in a plastic bag which fits tightly to the contents. Explain why atmospheric pressure is important in vacuum packaging.
- The Magdeburg hemispheres ($A = 1'170 \text{ cm}^2$) were pressed together with a force of 11.2 kN. The atmospheric pressure was 999 mbar.
 - What was the gauge pressure (difference in pressure between the inside of the sphere and atmospheric pressure)?
 - What was the pressure on the inside of the sphere?
- Fill a hose with water and close it at both ends with your thumbs. Then hold it as in the picture.
 - What happens if you open the lower end of the hose? Why?
 - What happens if you open the higher end of the hose (keeping the lower end open)? Why?
 - What would have happened on the moon (without atmospheric pressure)?



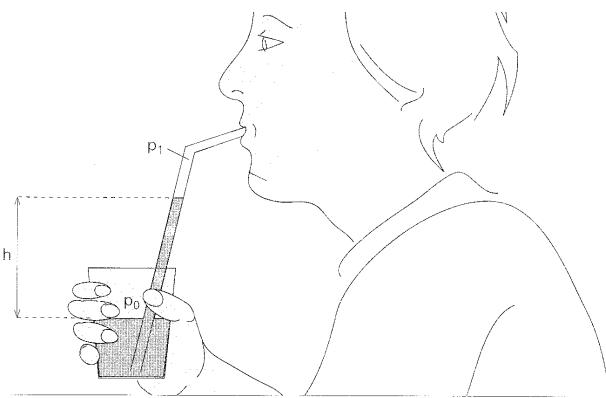
- 
 What is the height of the mercury column on Mount Everest ($p = 335 \text{ mbar}$)?



- A syringe is easily filled with water.
 - Explain the process without using the word *suction*.
 - Would it be possible to fill a syringe with water on the moon just as easily as on earth? Give reasons for your answer.

8. Drinking water through a straw at sea level.

- Explain the process without using the word *suction*.
- What is the atmospheric pressure p_0 at the surface of the water (outside the straw)?
- Calculate the fluid pressure at the surface of the water column inside the straw ($h = 20.0$ cm).
- At the water surface the pressure must be the same inside and outside the straw. The pressure of the water column and the pressure p_1 in the upper part of the straw add up to equal the atmospheric pressure p_0 .
What is the pressure p_1 ?
- What is the gauge pressure in the upper part of the straw? (Difference between atmospheric pressure and pressure inside the upper part of the straw)
- Would it be possible to drink through a straw on the moon? Give reasons for your answer.



solutions

- a) 1'013 mbar, 506.5 mbar, 253.3 mbar, 126.6 mbar, 63.3 mbar
d) ca. 750 mbar
b) 33'000 m above sea level
e) ca. 7'000 m above sea level
- a) 957 mbar
b) 42 mbar
- 25 cm
- b) 1.013 bar
c) 1'958 Pa
d) 99.3 kPa
e) -1'958 Pa